

MINIREVIEW

A PERSPECTIVE ON PHOTOSYNTHESIS IN THE OLIGOTROPHIC OCEANS: HYPOTHESES CONCERNING ALTERNATE ROUTES OF ELECTRON FLOW¹

Arthur R. Grossman²

Department of Plant Biology, Carnegie Institution of Science, 260 Panama Street, Stanford, California 94305, USA

Katherine R. M. Mackey

Civil and Environmental Engineering, Stanford University, Stanford, California 94305, USA

and Shaun Bailey

Department of Plant Biology, Carnegie Institution of Science, 260 Panama Street, Stanford, California 94305, USA

Many regions of the open, oligotrophic oceans are depleted of nutrients, especially nitrogen and iron. The biogenesis and the functioning of the photosynthetic apparatus may be specialized and tailored to the various marine habitats. In this mini-review, we discuss some new findings with respect to photosynthetic processes in the oceans. We focus on findings that suggest that some cyanobacteria may route electrons derived from the splitting of H₂O to the reduction of O₂ and H⁺ in a water-to-water cycle, and that other cyanobacteria that fix nitrogen during the day are likely missing PSII and enzymes involved in the fixation of inorganic carbon. Both of these proposed “variant” forms of photosynthetic electron flow provide new insights into ways in which marine phytoplankton satisfy their energetic and nutritive requirements.

Key index words: ecosystem; iron; nitrogen fixation; oceans; oligotrophic; oxidase; photosynthesis; photosystem

Abbreviations: C, carbon; cyt, cytochrome; DBMIB, 2,5-dibromo-3-methyl-6-isopropyl-p-benzoquinone; DCMU, 3-(3,4-dichloro-phenyl)-1,1-dimethyl-urea; ETC., electron transport chain; Fe, iron; F_m , maximal fluorescence; FNR, ferredoxin NADP-oxido-reductase, F_v/F_m , variable fluorescence divided by the maximal fluorescence, which is a measure of the dark-adapted photochemical efficiency of PSII; F_v , variable fluorescence; Fx, ferredoxin; N, nitrogen; P700, reaction center chl of PSI; PQ, plastoquinone; PS, photosystem; PTOX, plastoquinol terminal oxidase

The term oligotrophic comes from the Greek and literally means “small” or “little food”; nutrient resources can be very scarce in oligotrophic ecosystems. Oligotrophic oceans represent ~70% of the marine environment, are generally distant from coastal areas that provide nutrient inputs, and often have highly diminished iron (Fe) and nitrogen (N) levels (Zehr and Ward 2002, Arrigo 2005, Kupper et al. 2008). In spite of a “severe” nutrient-deprived lifestyle, picophytoplankton (photosynthetic phytoplankton of <2 μm) thrive in the oligotrophic ocean environment and may be responsible for 50% of the earth’s primary productivity (Field et al. 1998, Behrenfeld et al. 2006). Dominant picophytoplankton in oligotrophic oceans are represented by prokaryotic cyanobacteria of the genera *Prochlorococcus* and *Synechococcus*. *Prochlorococcus*, typically 0.5–1.0 μm in diameter with a divinyl chl-based light-harvesting complex, was discovered in 1988 (Chisholm et al. 1988). This organism is one of the most abundant bacteria on the planet and is detected at depths in the water column down to 200 m (Olson et al. 1990). *Synechococcus*, also small (usually 1.0–2.0 μm), has a distinctive light-harvesting complex composed mostly of pigmented phycobiliproteins. The oligotrophic environment also supports the growth of many eukaryotic algae including picoeukaryotes such as *Ostreococcus* spp. The major issue highlighted in this brief review focuses on novel, potentially critical aspects of photosynthesis performed by oceanic phytoplankton, and especially those in the oligotrophic oceans.

In oxygenic photosynthesis, energy from the sun is used to split water, extracting electrons and causing a light-driven vectorial electron flow that generates reductant and energy used for various processes, including the fixation of CO₂ into biomass. The complexes of the electron transport chain (ETC.) include the water-splitting complex,

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²Author for correspondence: e-mail arthurg@stanford.edu.

two photosystems (PS), PSI and PSII, that are excited by light energy absorbed by light-harvesting pigment protein complexes, the cytochrome (cyt) b_6f complex and an ATP synthase. The electron carriers that connect PSII to PSI are involved in pumping protons across the photosynthetic membranes; the proton gradient generated can drive the forma-

tion of ATP. A simplified schematic of photosynthetic electron transport is depicted in Figure 1A.

Interestingly, there are a number of intriguing features of photosynthesis that have been associated with the oligotrophic marine environment (Morel and Price 2003). Cyanobacteria and some eukaryotes that reside in the open oceans appear to have low

LOW RESOLUTION COLOR FIG

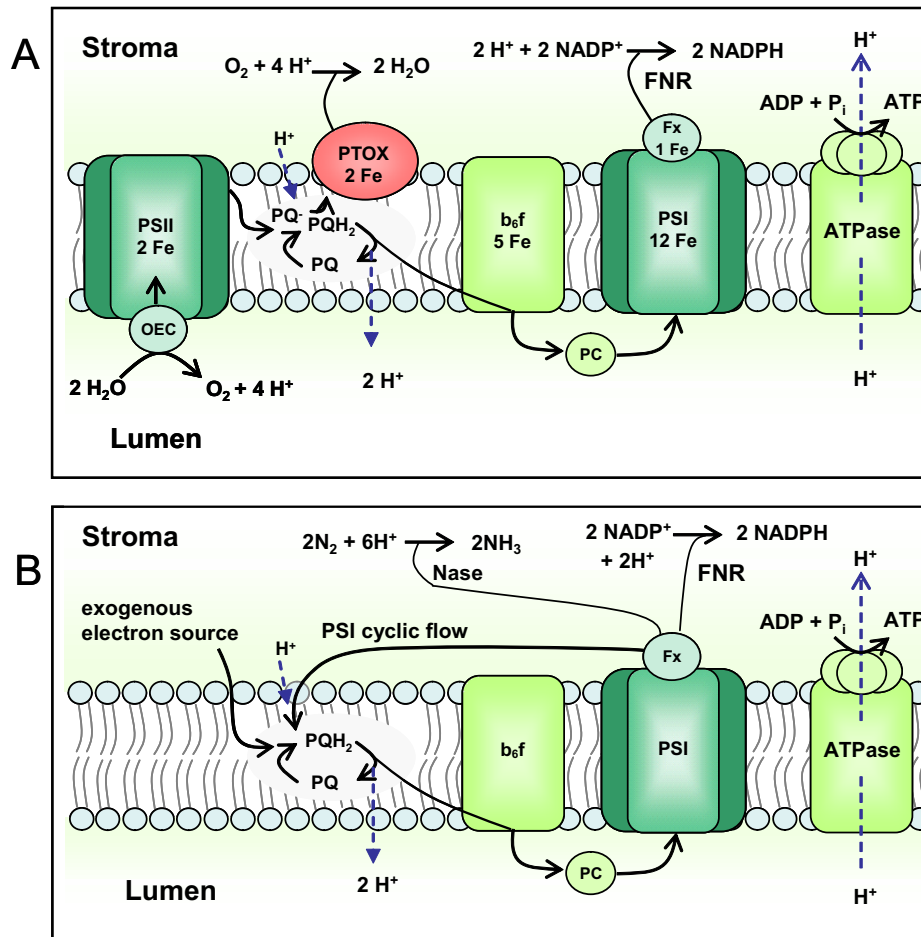


FIG. 1. (A) Photosynthetic electron transport—principle components, electron flow pathways, and iron requirements of some photosynthetic organisms in the oligotrophic oceans. This is a model proposed for the role of plastoquinone terminal oxidase (PTOX; or a similar oxidase) in photosynthetic electron flow associated with some marine phytoplankton in the oligotrophic oceans. Photosynthetic electron transport begins with excitation of the photosynthetic reaction centers, PSII and PSI, and water splitting by the oxygen-evolving complex (OEC) of PSII. Electrons are passed from PSII to the plastoquinone pool (to generate PQ^- and then, with the entry of protons from the stromal compartment, PQH_2). The electrons from PQH_2 are then sequentially transferred to cytochrome b_6f (b_6f), with a deposition of protons into the lumen of the thylakoid membranes, the mobile carrier plastocyanin (PC), PSI, ferredoxin (Fx), and then to ferredoxin NADP-oxido-reductase (FNR). FNR reduces $NADP^+$ and H^+ to form NADPH, which can be used for CO_2 fixation. In the PSII H_2O -to- H_2O cycle, electrons are extracted from the PQ pool (e.g., from reduced, unprotonated PQ^-), upstream of the Fe-rich cytochrome b_6f and PSI complexes, by a terminal oxidase (PTOX or another oxidase with some properties that are similar to that of PTOX) and that can reduce O_2 and H^+ to H_2O . ATP is formed by ATP synthase (ATPase) using the proton gradient (dotted blue arrow) generated either during traditional electron flow (cytochrome b_6f -mediated proton pumping) or via the H_2O -to- H_2O cycle (production of protons by the OEC in the lumen and consumption of protons by PTOX in the stroma). The number of iron (Fe) associated with each of the complexes is given below the name of each of the complexes. Black arrows indicate electron flow, while blue broken arrows indicate proton translocation. (B) Putative electron flow pathway in oceanic UCYN-A cyanobacteria lacking the PSII apparatus. It has been suggested that UCYN-A cells are photoheterotrophic, using PSI to generate ATP for nitrogen fixation but not to fix C by the Calvin–Benson–Bassham cycle. Electrons could be introduced into the electron transport chain via organic substrates, although the immediate source has not been identified. Electron flow may proceed as in the traditional linear electron flow pathway, terminating with the formation of reductant (NADPH) at the level of PSI, or may form a cycle around PSI, thereby facilitating ATP synthesis by helping to establish a proton gradient. Electrons from PSI may also proceed directly from ferredoxin (FX) to nitrogenase (Nase), where they reduce atmospheric N_2 to bioavailable NH_4^+ . Black arrows indicate electron flow, while blue broken arrows indicate proton translocation. PQ, plastoquinone.

levels of PSI and the cyt b_6f complex relative to PSII (Strzpek and Harrison 2004, Bailey et al. 2008, Cardol et al. 2008). This situation may have evolved since PSI and cyt b_6f are Fe-rich complexes with 12 and five atoms of Fe, respectively (Golbeck 2003). Since nutrient resources in oligotrophic oceans are scarce, and have been scarce for billions of years, some oligotrophic ocean organisms may not develop high levels of PSI even when supplemented with Fe (Bailey and Grossman 2008). Also, phytoplankton of oligotrophic waters show a peak value for dark-adapted photochemical efficiency of PSII (F_v/F_m) at dawn, and for photoinhibition (decreased F_v/F_m) at midday when irradiance is highest; these features are linked to low Fe and N availability (Behrenfeld et al. 2006). However, the photochemical efficiency of PSII in the light (Φ_{PSII}) remains remarkably stable at midday despite photoinhibition (Mackey et al. 2008), suggesting a mechanism to keep a population of PSII reaction centers or “traps” open (i.e., able to accept electrons) while electron input is high (e.g., in high light).

The observations described above may relate to the surprising finding that even when CO₂ fixation is light saturated, PSII photosynthetic electron transport can remain high; PSII traps can remain open at light intensities at which CO₂ fixation no longer increases. These findings were demonstrated in laboratory studies with the oligotrophic cyanobacterium *Synechococcus* WH8102, for which a significant proportion of PSII traps remained open at light intensities of close to 2,000 $\mu\text{mol photons} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$, even though CO₂ fixation saturated at $\sim 200 \mu\text{mol photons} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ (Bailey et al. 2008). A similar phenomenon occurs in the open-ocean picoeukaryote *Ostreococcus taurii* (Cardol et al. 2008), as well as in mixed populations of oligotrophic picophytoplankton (Mackey et al. 2008).

Studies discussed above raise important issues with respect to the ways in which photosynthetic electrons are used in both prokaryotic and eukaryotic marine phototrophs (Behrenfeld et al. 2008). A growing body of data suggests that at least some populations of marine phytoplankton have developed a means of extracting a significant proportion of electrons from PSII (either directly or indirectly) without using them for CO₂ fixation, raising a number of important questions: (i) Where are the extracted electrons going if they are not being used to fix CO₂? (ii) What mechanism(s) are involved in removal of these electrons? (iii) Are such activities advantageous in oligotrophic oceans, and if so how? (iv) Are these phenomena prevalent in nature? While many of these questions are still not fully answered, experimentally addressing these questions will lead to a better understanding of the energetics of photosynthesis and primary productivity in the oceans.

First, consider the relatively low PSI to PSII ratio in oligotrophic picophytoplankton. This feature can

result in an imbalance in electron flow through the two photosystems and cause a serious problem since PSI-dependent reoxidation of PSII might be slow, and excited singlet chl of PSII could accumulate leading to the production of potentially toxic levels of reactive oxygen species. This situation may be partially ameliorated by developing a larger PSI light-harvesting complex, which would facilitate more efficient PSI function at low light. This does occur in some Fe-limited cyanobacteria, with the large antennae formed by the IsiA protein (Nield et al. 2003). While the *isiA* gene does not appear to be prevalent in oligotrophic oceans, it is genetically diverse in coastal and high-nutrient, low chl marine ecosystems (Bibby et al. 2009, Rivers et al. 2009). **1** The strategy to generate a large antenna for PSI in Fe-poor environments might be beneficial at low light. However, in high light, it would favor highly reduced PSI reaction centers (P700), especially if the rate of downstream CO₂ fixation is low (e.g., due to low nutrient availability), and the P700 anion that accumulates could cause PSI damage and dysfunction. More work is needed to determine if the synthesis of IsiA or similar strategies for maximizing PSI function are used by phytoplankton in the oligotrophic ocean.

A potentially critical feature of open-ocean picophytoplankton is their ability to efficiently extract electrons from PSII without concomitant CO₂ fixation. PSII electron flow in *Synechococcus* WH8102 and *O. taurii* can be uncoupled from CO₂ fixation (Bailey et al. 2008, Cardol et al. 2008). For *Synechococcus* WH8102 in the light, PSII traps attained an open state as PSI became starved for electrons; a significant proportion of the electrons from PSII appeared to never reach PSI. Experiments designed to determine mechanism(s) required to maintain open PSII traps in high light demonstrated the dependence of this phenomenon on O₂; the traps closed rapidly when cells were placed under anaerobic conditions, but gradually opened as oxic conditions were reestablished in cultures (Bailey et al. 2008). Furthermore, inclusion of propyl gallate (pgal), an inhibitor of the plastoquinol terminal oxidase (PTOX), in the assays caused greater closure of PSII traps in the light (Bailey et al. 2008). A similar oxidase-dependent activity was observed in the open-ocean ecotype of *O. taurii* based on various observations. First, the quantum yield of PSII (Φ_{PSII}) was inhibited by pgal without altering the capacity of the cells for O₂ evolution, and a light-driven ΔpH was observed even when electron flow was blocked at the level of the cyt b_6f complex (by DBMIB). This “extra” ΔpH could be inhibited if photosynthetic electron transport was blocked at the level of PSII by DCMU, or if pgal was added to the reaction mixture (Cardol et al. 2008). When the results presented above are considered together, they strongly suggest that in at least some of the marine phytoplankton, electrons from PSII are **2**

being used to reduce O_2 on the stromal side of the photosynthetic membranes, at a site in the electron transport chain prior to cyt b_6f ; this would result in a H_2O -to- H_2O cycle for electron flow.

Light-stimulated O_2 uptake by phytoplankton has been discussed by others (Behrenfeld et al. 2008), and early studies by Falkowski demonstrated a light-enhanced rate of respiration in the marine diatom *Thalassiosira weissflogii* (Weger et al. 1989). Recently, light-stimulated O_2 uptake was examined for a variety of eukaryotic marine phototrophs (mostly coastal) (Suggett et al. 2009). Fast repetition rate fluorometry, mass inlet membrane spectrometry (MIMS), and ^{14}C uptake were used for simultaneous evaluation of PSII electron transport, gross and net O_2 evolution, and the fixation of inorganic carbon by six microalgal species (*Dunaliella tertiolecta*, *Pycnococcus provasoli*, *Storeatula major*, *Aureococcus anophagefferens*, *T. weissflogii*, and *Prorocentrum minimum*). Importantly, the MIMS data demonstrated that the slope of gross relative to net O_2 uptake exceeded 1 for three of the algal species examined (*P. minimum*, *P. provasoli*, and *S. major*), suggesting a significant light-elicited uptake of O_2 by these organisms (Suggett et al. 2009). Furthermore, light-stimulated O_2 uptake that likely occurs close to the site of photosynthetic O_2 evolution in a number of different phytoplankton was recently demonstrated by Luz and Kaplan through analyses of heavy O isotope fractionation (Eisenstadt et al. 2010). At this point, the molecular mechanisms associated with these O_2 -uptake activities have not been clearly defined, although they could involve the Mehler reaction, chlororespiration, cytochrome and alternative oxidase-dependent mitochondrial respiration, or pathways for alternative electron flow (e.g., PTOX-dependent O_2 uptake).

The discussion presented above suggests that a number of photosynthetic organisms that thrive in marine environments use O_2 as an electron acceptor and that this could potentially help maintain PSII traps in an open state when CO_2 fixation is saturated. Inhibition of electron extraction by pgal, at least for *Synechococcus* WH8102 and *O. taurii*, suggests that an oxidase related to PTOX, a quinol oxidase with homology to the mitochondrial alternative oxidase (Carol et al. 1999), can function to extract electrons from the photosynthetic ETC., most likely from the plastoquinone (PQ) pool. These conclusions are also supported by the finding that genes encoding PTOX are prevalent in metagenome databases generated from open-ocean samples (e.g., Sargasso Sea samples) (McDonald and Vanlerberghe 2005). Furthermore, the *ptox* gene has been found integrated into the genomes of marine cyanomyoviruses, which are related to the bacteriophage T4 (Millard et al. 2009), where it may impact viral-host interaction and cyanobacterial evolution. While others have shown that the photoreduction of O_2 can take place on the acceptor side (downstream) of

PSI through the Mehler reaction (Asada 1999), the extent to which such a reaction occurs on the donor side (upstream) of PSI in oceanic organisms such as *Synechococcus* WH8102 (Bailey et al. 2008), the picoeukaryote *O. taurii* (Cardol et al. 2008), and likely in assemblages of phytoplankton from the open ocean (Mackey et al. 2008) is surprisingly high (as much as 50%). The PSII H_2O -to- H_2O cycle, which may be a significant component of photosynthetic electron flow in marine phytoplankton, has been integrated into the photosynthetic electron transport scheme presented in Figure 1A. The potential functions of this cycle in organisms that thrive in Fe-poor oligotrophic oceans are as follows:

- 1 Reduce the need for high PSI levels, the most Fe-rich complex in the cell, since electrons could be extracted from the system prior to PSI. While the H_2O -to- H_2O cycle would limit CO_2 fixation, the growth potential of the cells would already be low because of low nutrient (e.g., N, Fe) levels.
- 2 When PSI levels are low or PSI activity is constrained, it is critical that the PSII traps be kept open to protect the system from photodamage. Creating an electron valve on the acceptor side (downstream) of PSII could allow for rapid extraction of electrons from intersystem electron transport under high-light conditions or when downstream CO_2 fixation is saturated. The redox pressure resulting from PSII activity may also be alleviated to some extent by non-photochemical quenching (Wilson et al. 2007).
- 3 Providing a strong electron outlet from the PQ pool would reduce electron flow to PSI and cyt b_6f , favoring a cationic P700 that may protect PSI reaction centers from photodamage, as previously suggested (Karapetyan 2007). This function of the PSII H_2O -to- H_2O cycle may be critical since PSI repair would require energy, and it may not be as easy to reassemble this reaction center when Fe levels are low.
- 4 The cycle may also serve a function similar to that of PSI cyclic electron flow. It would facilitate the generation of protons in the thylakoid lumen and the reduction of protons and O_2 on the cytosolic side of the thylakoid membranes, which would create a ΔpH for ATP synthesis, augmenting the cell's ability to efficiently scavenge ions from the environment.

Photosynthesis in some organisms in the oligotrophic oceans may have also become tailored for the fixation of N_2 , a process poisoned by O_2 . In oligotrophic regions, biological N_2 fixation has been traditionally attributed to the cyanobacterium *Trichodesmium* and cyanobacterial symbionts. However, recently it has been shown that there is a clade of unicellular cyanobacteria, "group A" or UCYN-A, that fix N_2 and that are related to *Cyanothece* sp. strain ATCC 51142, a marine unicellular

cyanobacterium isolated from an intertidal habitat, and to cyanobacteria that are symbiotic with the diatom *Rhopalodia gibba* (Precht et al. 2004). While these organisms have been found both in the Atlantic and widely distributed in the North and South Pacific oceans (Falcon et al. 2002, Zehr et al. 2008), they have not been successfully cultivated in the laboratory. New sequence information that contains a high degree of coverage of the UCYN-A genome suggests that these unicellular cyanobacteria have genes encoding PSI and nitrogenase (the N₂-fixation enzyme) but lack genes encoding components of the carbon concentration mechanism, the Calvin–Benson–Bassham Cycle and PSII (Zehr et al. 2008). These findings were supported by the inability of the researchers to amplify a PSII reaction center gene (*psbA*) and the gene encoding the LSU RUBISCO from UCYN-A-enriched DNA samples (in contrast, the genes encoding PSI components were readily amplified). These results suggest that the UCYN-A cyanobacteria perform cyclic electron flow around PSI (which would generate ATP) without evolving O₂ (depicted in Fig. 1B). Furthermore, this novel adaptation would explain how some cyanobacteria of the oligotrophic oceans can fix N₂ during the day when most diazotrophs (N₂ fixers) cannot. Rather than fixing N₂ within specialized, O₂-impermeant heterocyst cells, or at night when oxygenic photosynthesis ceases, the intracellular O₂ levels in the UCYN-A cyanobacteria may remain low enough during the day to allow for nitrogenase function. The lack of CO₂-fixation genes in these cyanobacteria suggests a photoheterotrophic mode of growth in which N₂ fixation is uncoupled from CO₂ fixation. Hence, achieving this highly specialized adaptation for N₂ fixation comes with the tradeoff that these organisms must exploit other sources of organic carbon/reductant for both growth and nitrogen fixation (indicated as “exogenous electron source” in Fig. 1B), and they must also efficiently scavenge Fe from the environment to produce PSI centers and Fe-rich nitrogenase enzymes. The levels, activity, and composition of PSI in these cyanobacteria need to be carefully characterized. These findings are also likely to have significant implications with respect to the N and C budgets of the oligotrophic ocean.

In conclusion, the photosynthetic “variant processes” discussed above suggest a diversity of photosynthetic mechanisms that have evolved in phytoplankton of the oligotrophic ocean. A PSII H₂O-to-H₂O cycle appears to be a significant alternative route of electron flow in these organisms (and potentially in some organisms that grow in coastal regions). Interestingly, while an activity involving PTOX has been known to occur in plants for over a decade, and has been attributed roles in both carotenoid biosynthesis and poisoning of the redox state of the PQ pool (reviewed in Rumeau et al. 2007), it has recently become clear that it can

represent a more prevalent pathway in plants exposed to stress conditions (Quiles 2006), as well as in certain organisms adapted to survival in high light, low temperature (Streb et al. 2005), and high salt (Stepien and Johnson 2009) environments. We still need a clearer picture of the physiological/ecological relevance of the PSII H₂O-to-H₂O cycle in open-ocean organisms, how widespread the process is among the various phytoplankton (and if it does occur to any extent in organisms from coastal habitats), and the advantages/disadvantages associated with the various alternate routes of photosynthetic electron flow. It is also critical to understand how phytoplankton that synthesize large PSI antennae under Fe-limiting conditions, and others that may fix N₂ without oxygenic photosynthesis, impact C fixation and sequestration in the oceanic environment. Clearly, these variant processes have the potential to modify our understanding of the relationship between chl levels and primary productivity in oligotrophic environments. Evolution has tailored photosynthesis in the oligotrophic oceans over the course of many millions of years in ways that we are only just beginning to understand. This understanding is critical if we are to evaluate how photosynthetic processes and C cycling in the oceans will be impacted by the very rapidly changing environment that is being shaped by humans.

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Q2	AUTHOR: Cardol et al. 2009 has been changed to Cardol et al. 2008 so that this citation matches the Reference List. Please confirm that this is correct.	
Q3	AUTHOR: Sugget et al. 2009 has been changed to Suggestt et al. 2009 so that this citation matches the Reference List. Please confirm that this is correct.	
Q4	AUTHOR: Please check the usage 'irons'.	
Q5	AUTHOR: Figure 1 has been saved at a low resolution of 170 dpi. Please resupply at 600 dpi. Check required artwork specifications at http://authorservices.wiley.com/submit_illustrations.asp?site=1.	

Proof Correction Marks

Please correct and return your proofs using the proof correction marks below. For a more detailed look at using these marks please reference the most recent edition of The Chicago Manual of Style and visit them on the Web at: <http://www.chicagomanualofstyle.org/home.html>

<i>Instruction to typesetter</i>	<i>Textual mark</i>	<i>Marginal mark</i>
Leave unchanged	... under matter to remain	<u>stet</u>
Insert in text the matter indicated in the margin	^	^ followed by new matter
Delete	ƒ through single character, rule or underline or ƒ through all characters to be deleted	ƒ
Substitute character or substitute part of one or more word(s)	ƒ through letter or — through characters	new character ƒ or new characters ƒ
Change to italics	— under matter to be changed	<u>ital</u>
Change to capitals	≡ under matter to be changed	<u>Caps</u>
Change to small capitals	≡ under matter to be changed	<u>sc</u>
Change to bold type	~ under matter to be changed	<u>bf</u>
Change to bold italic	~ under matter to be changed	<u>bf+ital</u>
Change to lower case	ƒ	<u>lc</u>
Insert superscript	√	√ under character e.g. √
Insert subscript	^	^ over character e.g. ^
Insert full stop	⊙	⊙
Insert comma	↕	↕
Insert single quotation marks	↕ ↕	↕ ↕
Insert double quotation marks	↗ ↘	↗ ↘
Insert hyphen	=	=
Start new paragraph	¶	¶
Transpose	┌┐	┌┐
Close up	linking  characters	
Insert or substitute space between characters or words	#	#
Reduce space between characters or words	˘	˘